



a place of mind

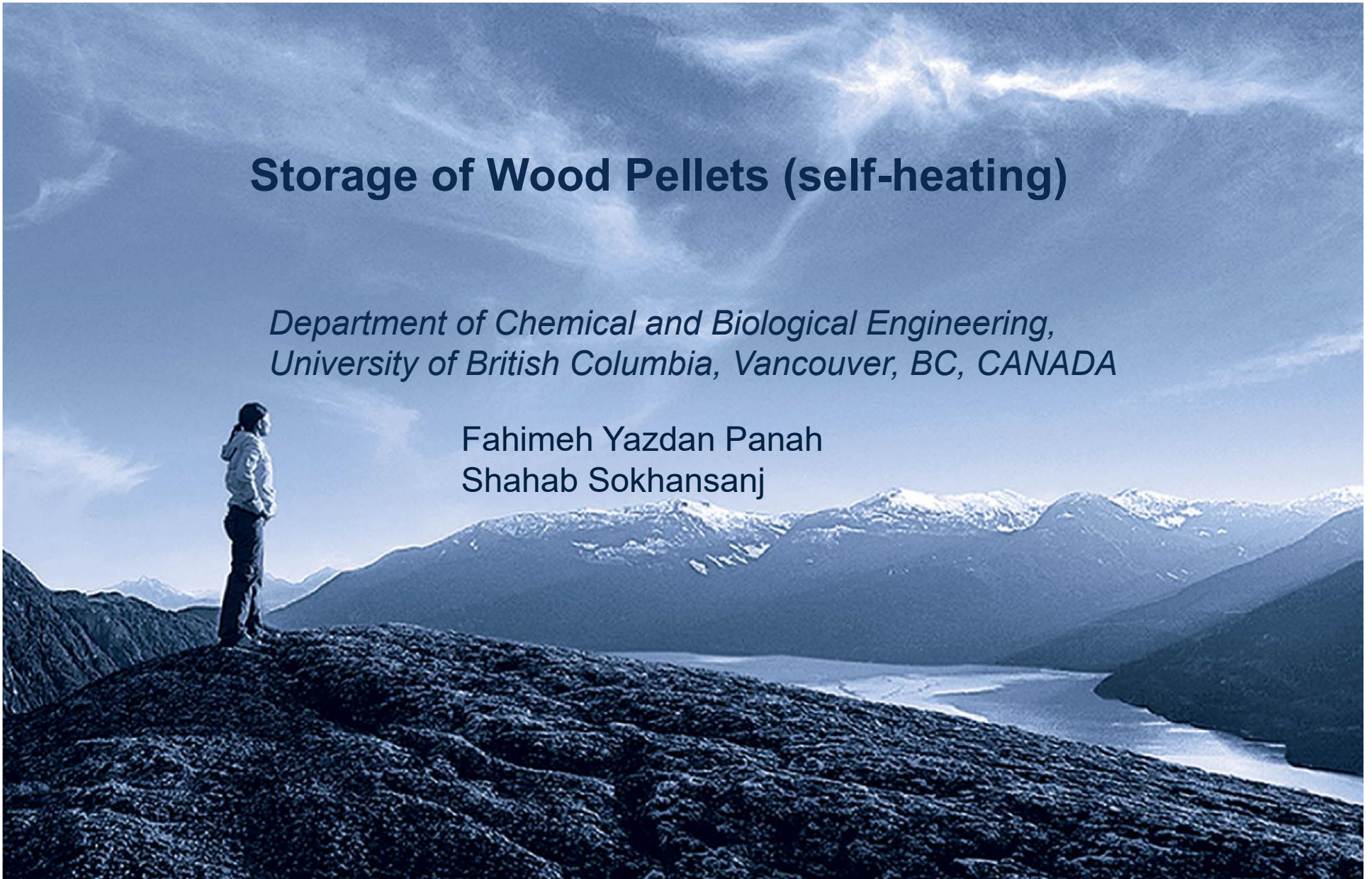
THE UNIVERSITY OF BRITISH COLUMBIA



Storage of Wood Pellets (self-heating)

*Department of Chemical and Biological Engineering,
University of British Columbia, Vancouver, BC, CANADA*

Fahimeh Yazdan Panah
Shahab Sokhansanj





Self-heating and thermal runaway

Thermal runaway Factors that

Smoldering combustion•

Chemical degradation •

Chemical oxidation•

Pyrolysiscontent.

Hydrolysis•

Biological

Metabolism and biological oxidation



influence self-heat are:

Oxygen concentration in the bulk

- Moistening of bulk

Relative humidity

Aggravating factors are:

Pellet temperature and moisture

Conduction and convection of heat and

degradation moisture in the pile.

- Physical properties (broken pellets and dust).



Self-heating studies and approaches

To quantify and predict self-heating process, runaway condition and spontaneous combustion of wood pellet during storage.

Previous Studies

Measurement of thermal properties.

- Thermal conductivity ○
- Specific heat capacity

Investigation of the thermal runaway .

- Self-heating kinetics ○

Thermal runaway critical condition

Theoretical model ○ 2 Phase heat /mass transfer ○ 2D axis symmetric model

Current Studies

Measurement of self-heating using Calorimetry.

- Moisture content effect ○
- Age effect ○ Feedstock effect
- Humidity, temperature



a place of mind

THE UNIVERSITY OF BRITISH COLUMBIA



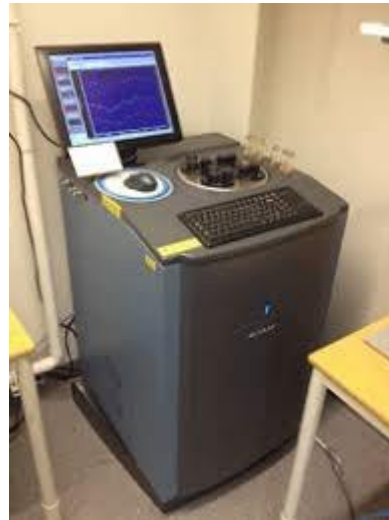
effect ○ Particle size effect ○
Storage time effect

Storage Studies Facilities at UBC



a place of mind

THE UNIVERSITY OF BRITISH COLUMBIA





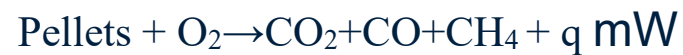
a place of mind

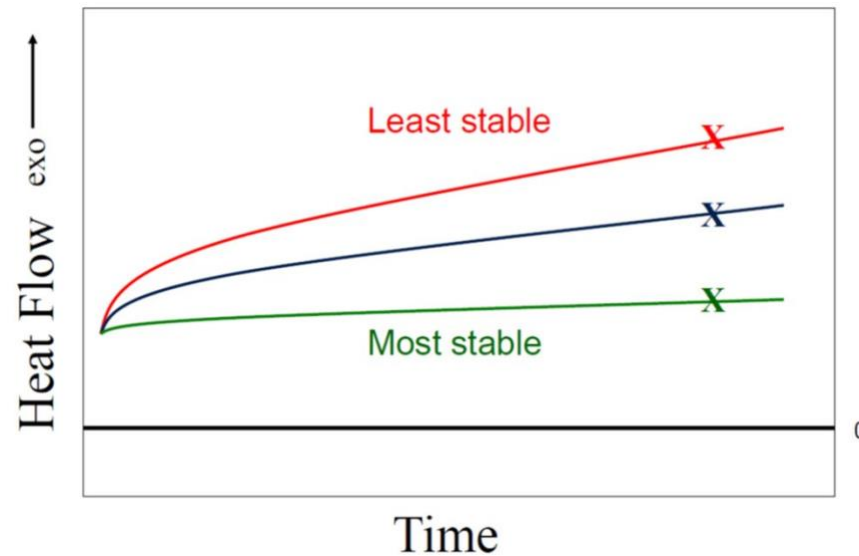
THE UNIVERSITY OF BRITISH COLUMBIA



Measurement of self-heating using Isothermal Calorimetry

- Self heating tests at 25/40/60°C

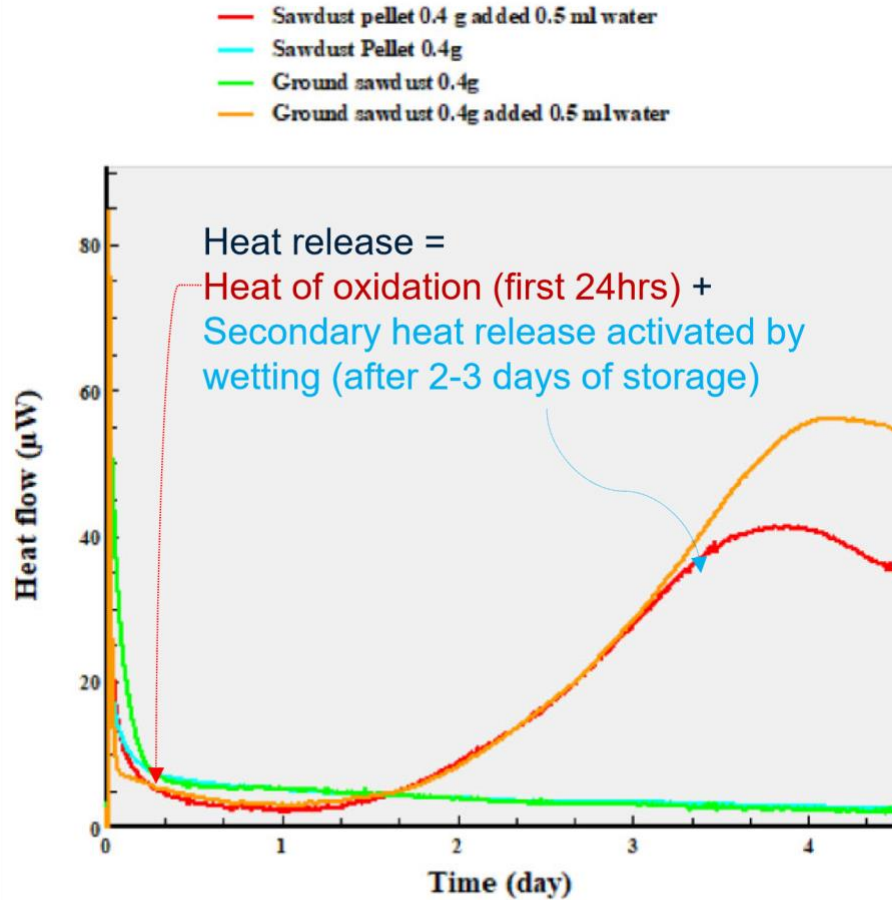




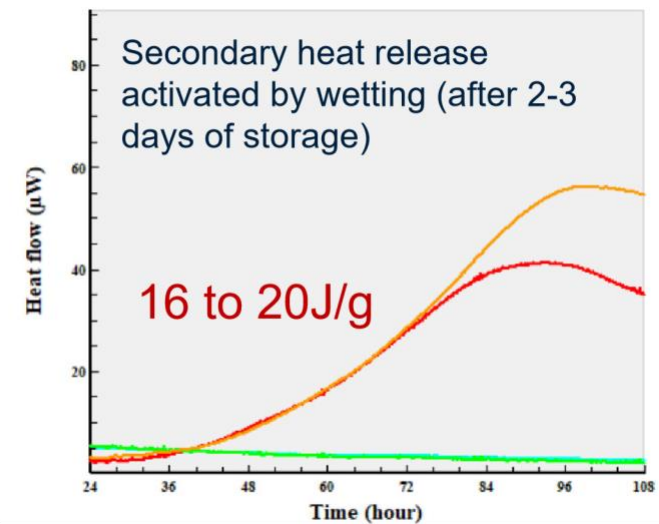
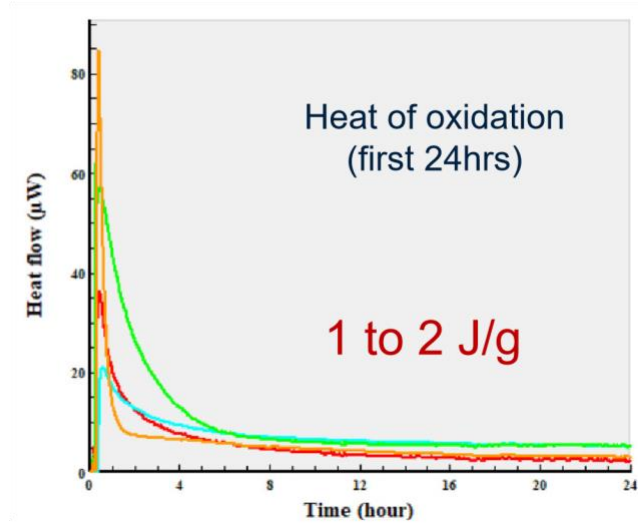
TAM III, TA Instrument

Calorimetry is sensitive to all physical and chemical processes associated with a heat flow. Thus, the monitored heat flow may contain contributions from several processes.

Isothermal Calorimetry- Effect of Moisture

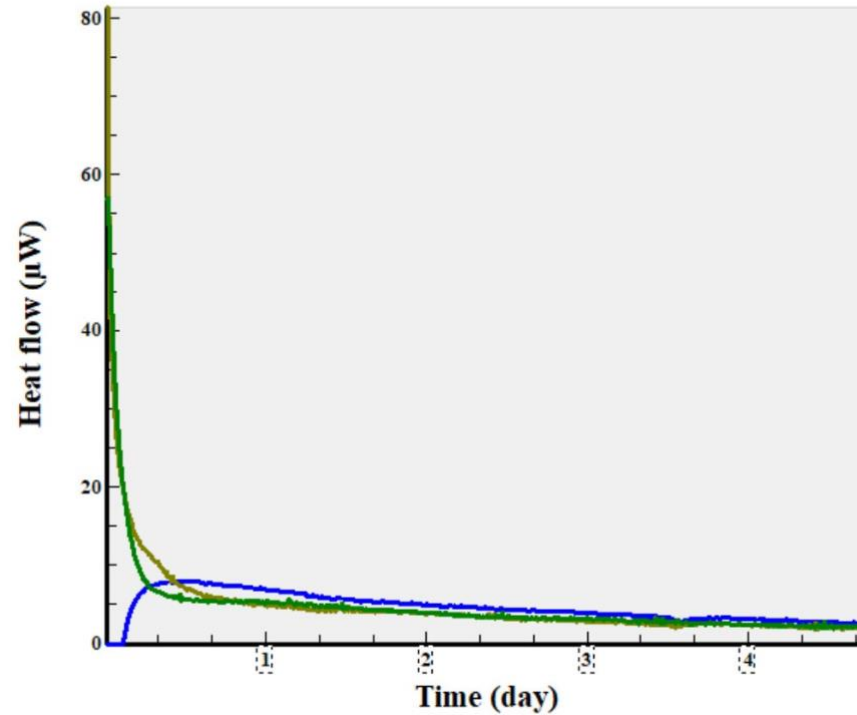


Heat released from ground sawdust over 5 day of storage in air with and without addition of moisture



Isothermal Calorimetry- Effect of Oxygen

- Ground sawdust 0.4g - O₂ Rich Environment
- Ground sawdust 0.4g - N₂ Rich Environment
- Ground sawdust 0.4g in air



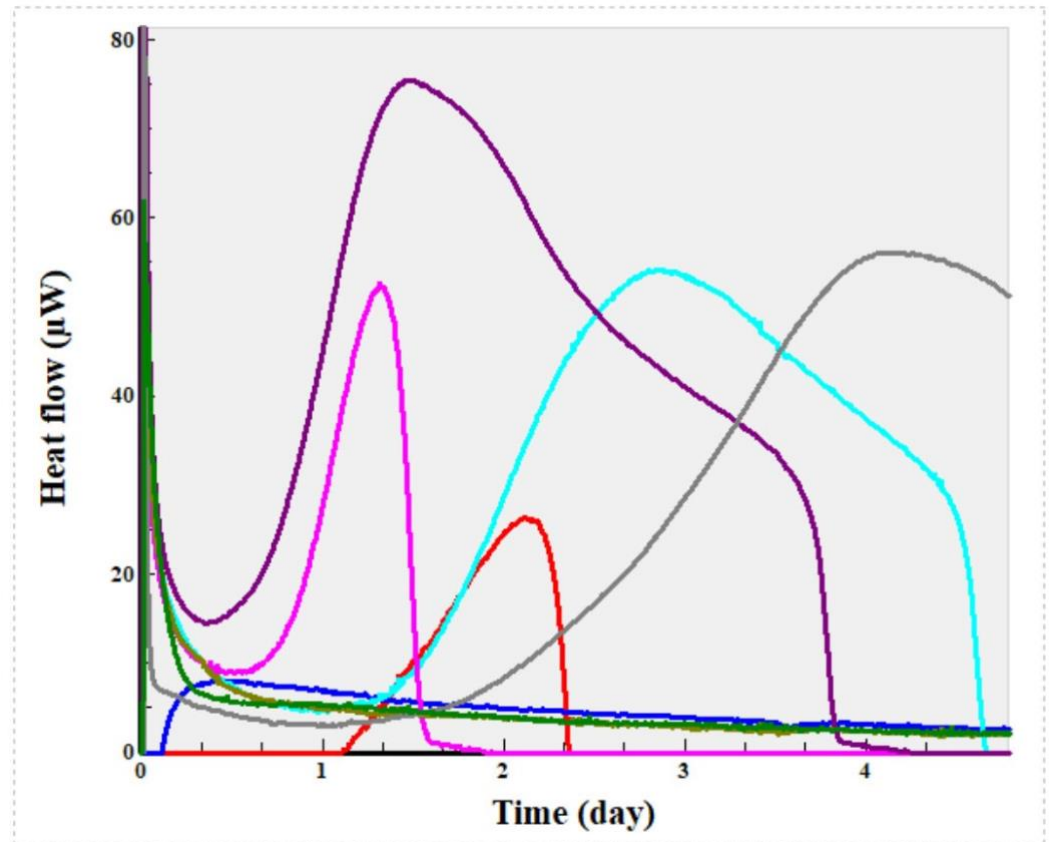
Effect of O₂ on heat release is minimal if not accompanied by wetting (moisture/humidity)

Heat released from ground sawdust over 5 day of storage in air, oxygen rich and nitrogen rich environment

Isothermal Calorimetry- Effect of Oxygen accompanied by wetting

- Ground sawdust 0.4g purged with N₂ added 0.5 ml water
- Ground sawdust 0.4g purged with O₂
- Ground sawdust 0.4g purged with O₂ added 1 ml water
- Ground sawdust 0.4g purged with N₂ added 1 ml water
- Ground sawdust 0.4g Purged with N₂
- Ground sawdust 0.4g purged with O₂ added 0.5 ml water
- Ground sawdust 0.4g added 0.5 ml water
- Ground sawdust 0.4g

- Significant effect of O₂ when accompanied by moisture is shown → auto-oxidation and thus significant heat release

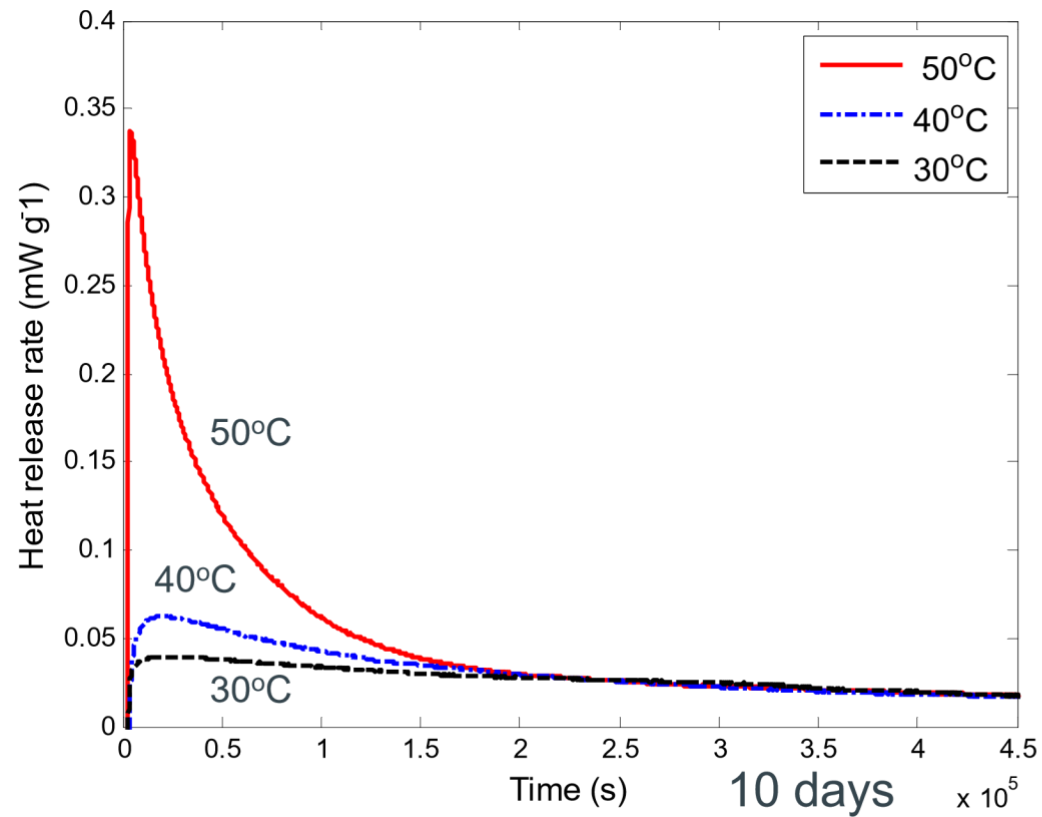




- In oxygen poor environment the reaction is not self-sustained → no auto oxidation
- The effect of wetting in heat release is significantly higher than oxygen. e.g. the heat

Heat released from ground sawdust over 5 day of
release in air & moist storage in air, oxygen rich and nitrogen rich environment is as
big as oxygen environment accompanied by wetting rich moist environment.

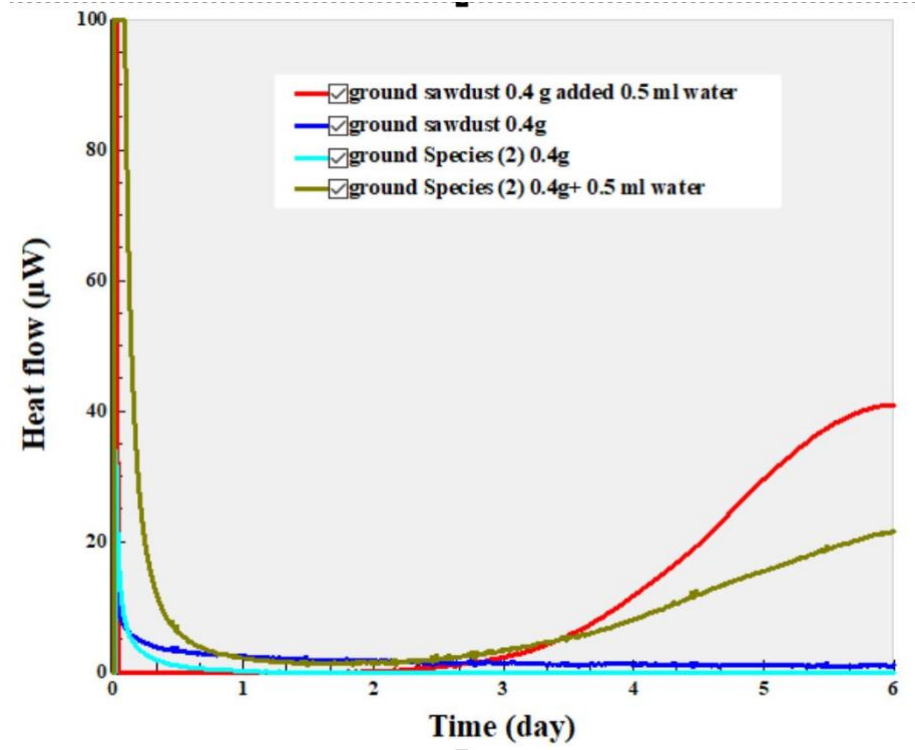
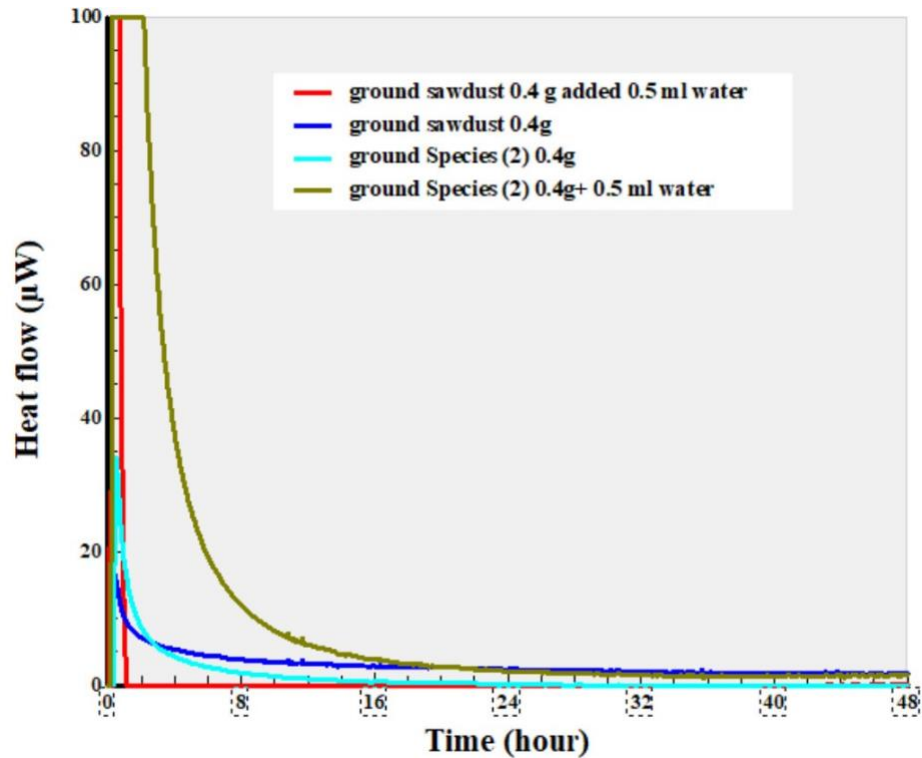
Effect of Storage Temperature



Temperature significantly affects the heat release rate

Heat released from sawdust over 10 days of storage in air stored at 30, 40 and 50C

Isothermal Calorimetry- Effect of species



Heat released from ground sawdust and another species over 6 days of storage in air (with and without moisture)



Lessons Learned

- The lower the storage temperature, the lower the risk of self-heating. Higher temperature increases both the initial heat release due to oxidation as well as the secondary auto-oxidation activated by wetting (humidity/moisture/rain)
- Amongst the species tested so far, sawdust has shown higher heat release when exposed to moist environment. Keep a balanced species mix.
- The effect of oxygen is minimal if not accompanied by moisture. The primary heat due to oxidation is slightly higher when pellets are stored in oxygen-rich environment. However this slight increase is insignificant when compared to heat release when the two parameters are of moisture and oxygen are present. Keep the material in low humidity away from rain and moisture.
- The secondary heat release of auto-oxidation that is activated by wetting (humidity, rain etc) is significantly higher (15-18 times more) compared to the primary heat release during storage.



Lessons Learned

- Rank and change in moisture are judged to be the most important factors affecting the spontaneous heating of the pile
- The two processes, oxidation and wetting contribute to spontaneous heating. The amount of energy released by oxidation and /or wetting determines the spontaneous heat hazard.
- A more general trigger of sustained, self-accelerating oxidation is seen in heat releases that accompany wetting of partly dry material. When the pellet is drier, more heat can be released from water adsorption (less heat is needed for desorption)
- The rate of heat generation is finely balanced with the rate of heat loss. **The humidity of the air is an important factor in deciding whether a heating will progress rapidly or not**



a place of mind

THE UNIVERSITY OF BRITISH COLUMBIA



Acknowledgements

- Wood Pellets Association Canada (WPAC)
- Natural Sciences and Engineering Research Council of Canada (NSERC)
- Biomass and Bioenergy Research Group



a place of mind

THE UNIVERSITY OF BRITISH COLUMBIA



Thank You

